CHM 111

College Chemistry II

Course Description
This course is a continuation of the study of atomic and molecular structure, nomenclature and equations, properties, reactions and states of matter, stoichiometry, gas laws, solutions, and equilibria. Other topics included are kinetics, thermodynamics, and electrochemistry.

Prerequisites: CHM 110

4.0 Cr (3.0 lect/pres, 3.0 lab, 0 other)

Course Focus
Understand the structure, nomenclature and reactions of various organic compounds; name and write structural formulas for the following classes of organic chemicals and will learn the typical chemical reactions and physical properties there of: alkanes, alkenes, alcohols, alkyl halides, aromatic compounds, amines, aldehydes, ketones, and carboxylic acids and their derivatives; distinguish between alkanes, alcohols, and aldehydes/ketones, Understand the chemistry of carbohydrates, their polymers and their properties, Distinguish the characteristics and properties of carboxylic acids and esters, Understand the various classes and properties of lipids and their role in biology, acquire a fundamental understanding of amino acids and proteins, the structures of proteins and their function in metabolism, understand the basic structure of nucleic acids, distinguish between DNA and RNA and define the role of nucleic acids in transmitting genetic information

Text and References

CHM 111 CORE CURRICULUM COMPETENCIES

All courses approved for the general education core curriculum help students develop communication skills and/or critical thinking. Students will demonstrate achievements by assessments on the departmental final exam and on testing developed by individual instructors. This course develops critical thinking skills through instruction that emphasizes the understanding of the scientific disciplines of general chemistry, as demonstrated in the following: a formal research paper requiring the student to develop, evaluate, and synthesize credible information on a given topic. This research project allows the student to:

- Apply standard scientific methods and interpret laboratory observations and data;
- Make inferences justified by data and observations;
- Explain relevance of findings to general chemical principles, or expected results;
- Identify key assumptions of general chemistry.

Periodic exams, homework and a standardized final examination also tests critical thinking ability.

Course Goals
The following list of course goals will be addressed in the course. These goals are directly related to the performance objectives. (*designates a CRUCIAL goal)

1. Classify organic molecules *
2. Identify organic properties
3. Write iupac names
4. Give common names *
5. Name alkanes *
6. Draw structural formulas
7. Draw polymer formulas
8. Identify monomer components *
9. Illustrate hydrocarbon structures
10. Classify aldehydes and ketones
11. Identify aldehydes and ketones
12. Classify carboxylic acids
13. Name carboxylic acids
14. Name esters
15. Form esters
16. Classify amines and amides
17. Identify amines and amides
18. Classify alcohols
19. Demonstrate dehydration synthesis *
20. Name aromatic compounds
21. Describe benzene bonding
22. Write alcohol reactions *
23. Write thiol reactions
24. List organic redox products
25. Name alcohols
26. Name ethers
27. Classify carbohydrates *
28. Draw common carbohydrate structures
29. List common carbohydrate examples
30. Build polysaccharides
31. Define glycosidic bond
32. Draw Hayworth structures
33. Predict lipid behavior
34. Draw cis-trans isomers *
35. Distinguish saturated and unsaturated fatty acids *
36. Identify chiral carbon
37. Predict hydrogenation products *
38. Understand hydrocarbon addition reactions
39. Define lipid properties
40. Draw amino acids
41. Identify quaternary protein structure
42. Describe protein function
43. Assemble nucleotides *
44. Distinguish purine from pyrimidine
45. Distinguish DNA and RNA
46. Draw DNA structure
47. Explain enzyme catalysis *
48. Explain enzyme inhibition
49. Trace protein synthesis *

Student Contributions
Classes are designed to employ a variety of teaching techniques. In order to maximize learning, required readings should be done prior to a unit. If a student is falling behind in lab performance or academic achievement, it is imperative to seek immediate assistance from the instructors.

Course Evaluation
- Student progress will be evaluated through a series of tests, quizzes in-class and out of class assignments and will be detailed in the attachment to this syllabus.
- Blackboard: lecture notes, handouts, podcasts, study hints, tutor information, syllabi, and other course information is available on the course blackboard page.
- Laboratory Component: This course has a required lab component which supplements the information presented in lecture. The lab will be independently evaluated primarily through lab practicals, in class and out of class lab assignments (such as research papers). For specific details about lab evaluations, please refer to the attachment to this syllabus.

GRADING SCALE:

90-100 = A
80-89 = B
70-79 = C
60-69 = D
Below 60= F

Course Schedule
This course meets 2.5 hours per week for lecture, has a 0.5hr per week online component and 3.0 hours per week for lab.

**ADA STATEMENT**

The Technical College of the Lowcountry provides access, equal opportunity and reasonable accommodation in its services, programs, activities, education and employment for individuals with disabilities. To request disability accommodation, contact the counselor for students with disabilities at (843) 525-8228 during the first ten business days of the academic term.

**ACADEMIC MISCONDUCT**

There is no tolerance at TCL for academic dishonesty and misconduct. The College expects all students to conduct themselves with dignity and to maintain high standards of responsible citizenship.

It is the student’s responsibility to address any questions regarding what might constitute academic misconduct to the course instructor for further clarification.

The College adheres to the Student Code for the South Carolina Technical College System. Copies of the Student Code and Grievance Procedure are provided in the *TCL Student Handbook*, the Division Office, and the Learning Resources Center.

**ATTENDANCE**

The College’s statement of policy indicates that students must attend ninety percent of total class hours or they will be in violation of the attendance policy.

- Students not physically attending class during the first ten calendar days from the start of the semester must be dropped from the class for NOT ATTENDING.
- Students taking an online/internet class must sign in and communicate with the instructor within the first ten calendar days from the start of the semester to indicate attendance in the class. Students not attending class during the first ten calendar days from the start of the semester must be dropped from the class for NOT ATTENDING.
- Reinstatement requires the signature of the division dean.
- In the event it becomes necessary for a student to withdraw from the course OR if a student stops attending class, it is the student’s responsibility to initiate and complete the necessary paperwork. Withdrawing from class may have consequences associated with financial aid and time to completion.
- When a student exceeds the allowed absences; the student is in violation of the attendance policy. The instructor MUST withdraw the student with a grade of “W”, “WP”, or “WF” depending on the date the student exceeded the allowed absences and the student’s progress up to the last date of attendance or
- Under extenuating circumstances and at the discretion of the faculty member teaching the class, allow the student to continue in the class and make-up the work. This exception must be documented at the time the allowed absences are exceeded.
- Absences are counted from the first day of class. There are no "excused" absences. All absences are counted, regardless of the reason for the absence.
- A student must take the final exam or be excused from the final exam in order to earn a non-withdrawal grade.
- A copy of TCL’s STATEMENT OF POLICY NUMBER: 3-1-307 CLASS ATTENDANCE (WITHDRAWAL) is on file in the Division Office and in the Learning Resources Center.
HAZARDOUS WEATHER
In case weather conditions are so severe that operation of the College may clearly pose a hardship on students and staff traveling to the College, notification of closing will be made through the following radio and television stations: WYKZ 98.7, WGCO 98.3, WGZO 103.1, WFXH 106.1, WWVV 106.9, WLOW 107.9, WGZR 104.9, WFXH 1130 AM, WLVH 101.1, WSOK 1230 AM, WAEV 97.3, WTOR TV, WTGS TV, WJWJ TV, and WSAV TV. Students, faculty and staff are highly encouraged to opt in to the Emergency Text Message Alert System. www.tcl.edu/textalert.asp

EMERGENCY TEXT MESSAGE ALERT
Students, faculty and staff are highly encouraged to opt in to the Emergency Text Message Alert System. Participants receive immediate notification of emergency events and weather cancelations via text messaging on their cell phones. Participants can also opt in to receive non-emergency news and announcements. Go to www.tcl.edu. On the homepage, click on “emergency TextAlert at TCL” and fill out the form or go to www.tcl.edu/textalert.asp

GRADING METHODOLOGY
The final grade must be 70 or more (a grade “C” or better) in order to pass the course and progress to the next course. Students absent from an examination or presentation will receive a “0” grade for the examination unless other arrangements are made with the individual instructor prior to the examination or presentation day or on the examination or presentation day before the test/presentation is scheduled to be given.

The student is responsible for notifying the instructor for the reason of the absence. It is also the responsibility of the student to contact the appropriate instructor to arrange to make up the examination. Arrangements may be completed by telephone.

If the instructor is not available, a message should be left on the instructor’s voice mail AND with another member of the faculty or administrative assistant. The make-up exam will be scheduled and the instructor will decide the method of examination. Messages sent by other students are unacceptable.